



# A Glycolipid Glycosyltransferase with Broad Substrate Specificity from the Marine Bacterium “*Candidatus Pelagibacter* sp.” Strain HTCC7211

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**ABSTRACT** In the marine environment, phosphorus availability significantly affects the lipid composition in many cosmopolitan marine heterotrophic bacteria, including members of the SAR11 clade and the *Roseobacter* clade. Under phosphorus stress conditions, nonphosphorus sugar-containing glycolipids are substitutes for phospholipids in these bacteria. Although these glycolipids play an important role as surrogates for phospholipids under phosphate deprivation, glycolipid synthases in marine microbes are poorly studied. In the present study, we biochemically characterized a glycolipid glycosyltransferase (GT<sub>cp</sub>) from the marine bacterium “*Candidatus Pelagibacter* sp.” strain HTCC7211, a member of the SAR11 clade. Our results showed that GT<sub>cp</sub> is able to act as a multifunctional enzyme by synthesizing different glycolipids with UDP-glucose, UDP-galactose, or UDP-glucuronic acid as sugar donors and diacylglycerol (DAG) as the acceptor. Analyses of enzyme kinetic parameters demonstrated that Mg<sup>2+</sup> notably changes the enzyme’s affinity for UDP-glucose, which improves its catalytic efficiency. Homology modeling and mutational analyses revealed binding sites for the sugar donor and the diacylglycerol lipid acceptor, which provided insights into the retaining mechanism of GT<sub>cp</sub> with its GT-B fold. A phylogenetic analysis showed that GT<sub>cp</sub> and its homologs form a group in the GT4 glycosyltransferase family. These results not only provide new insights into the glycolipid synthesis mechanism in lipid remodeling but also describe an efficient enzymatic tool for the future synthesis of bioactive molecules.

**IMPORTANCE** The bilayer formed by membrane lipids serves as the containment unit for living microbial cells. In the marine environment, it has been firmly established that phytoplankton and heterotrophic bacteria can replace phospholipids with non-phosphorus sugar-containing glycolipids in response to phosphorus limitation. However, little is known about how these glycolipids are synthesized. Here, we determined the biochemical characteristics of a glycolipid glycosyltransferase (GT<sub>cp</sub>) from the marine bacterium “*Candidatus Pelagibacter* sp.” strain HTCC7211. GT<sub>cp</sub> and its homologs form a group in the GT4 glycosyltransferase family and can synthesize neutral glycolipids (monoglucosyl-1,2-diacyl-*sn*-glycerol [MGLc-DAG] and monogalactosyl [MGal]-DAG) and monoglucuronic acid diacylglycerol (MGLcA-DAG). We also uncovered the key residues for DAG binding through molecular docking, site-direct mutagenesis, and subsequent enzyme activity assays. Our data provide new insights into the glycolipid synthesis mechanism in lipid remodeling.

**KEYWORDS** lipid remodeling, SAR11, marine bacteria

Phospholipids form the structural basis of all cells, but sugar-containing glycolipids are mainly restricted to marine microbes, cyanobacteria, and higher plants (1, 2). Glycolipids are found on the lipid bilayer of cell membranes and play

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critical roles in cell growth, cellular recognition, adhesion, neuronal repair, and signal transduction. These natural glycosylglycerolipids often have unusual and sometimes unexpected biological activities, such as antitumor, antiviral, anti-inflammatory, antimalarial, immunostimulatory, and neuritogenic activities, which make them valuable molecular targets for research (3–5). The basic structure of glycosylglycerolipids is characterized by a 1,2-diacyl-*sn*-glycerol moiety with different numbers and types of sugars (glucose, galactose, mannose, rhamnose, or charged sugars like glucuronic acid or sulfoquinovose) attached at the *sn*-3 position of the glycerol backbone in diacylglycerol (DAG). These sugar attachments have an  $\alpha$ - or  $\beta$ -anomeric configuration and are bound via (1 $\rightarrow$ 2), (1 $\rightarrow$ 3), (1 $\rightarrow$ 4), or (1 $\rightarrow$ 6) linkages (2, 6). The common glycosylglycerolipid structures in marine heterotrophic microbes and cyanobacteria are 1,2-diacyl-3-*O*-( $\beta$ -D-galactopyranosyl)-*sn*-glycerol (monogalactosyl diacylglycerol [MGal-DAG]), 1,2-diacyl-3-*O*-( $\alpha$ -D-glucopyranosyl)-*sn*-glycerol (monoglucosyl DAG [MGlc-DAG]), 1,2-diacyl-3-*O*-( $\alpha$ -D-galactopyranosyl)-(1 $\rightarrow$ 6)-*O*-( $\beta$ -D-galactopyranosyl)-*sn*-glycerol (digalactosyl DAG [DGal-DAG]), and 1,2-diacyl-3-*O*-(6-deoxy-6-sulfo- $\alpha$ -D-galactopyranosyl)-*sn*-glycerol (sulfoquinovosyl DAG [SQDG]) (2, 7).

Glycosylglycerolipids are usually synthesized by glycosyltransferases (GTs), which are highly divergent and polyphyletic. The GTs can be categorized into 110 numbered families according to their sequence similarity and signature motifs and the stereochemistry of the glycoside linkage formed (8). Of the 110 families, the families GT4, GT21, and GT28, which are known as glycosylglycerolipid synthases, utilize sugar nucleotides as donors and contain a consensus sugar donor binding domain near the C terminus (9, 10). Despite the wide variety of bacterial glycosylglycerolipids, only a few bacterial lipid GTs have been identified and characterized so far. The GTs synthesizing MGlc-DAG and DGal-DAG have been isolated from the cell wall-less bacterium *Acholeplasma laidlawii* and were found to belong to the GT4 family in the carbohydrate-active enzymes (CAZy) database (11). Other known members of bacterial GT4 include the MGlc-DAG synthases from *Deinococcus radiodurans* and *Thermotoga maritima* and the MGal-DAG synthase from *Borrelia burgdorferi*, which was the first cloned galactosyltransferase forming MGal-DAG with the  $\alpha$ -anomeric configuration of the sugar (12, 13). A bifunctional GT (designated Agt) from *Agrobacterium tumefaciens* was found to synthesize MGlc-DAG or monoglucuronic acid diacylglycerol (MGlcA-DAG) with UDP-glucose (UDP-Glc) or UDP-glucuronic acid (UDP-GlcA) as the sugar donor, respectively (6). This enzyme also belongs to the GT4 family and was the first glucuronosyl DAG synthase to be isolated. The processive GTs (Pgts), however, are members of the GT21 family and show high sequence similarity to GT4 family GTs. The Pgts include enzymes from *Mesorhizobium loti* and *A. tumefaciens* that synthesize DGal-DAG, glucosylgalactosyl-DAG (GlcGal-DAG), and triglycosyl DAGs (14, 15). To the best of our knowledge, the structure-function relationship of DAG-dependent GT4 glycosyltransferases has not been studied previously due to the lack of a crystal structure of these glycosylglycerolipid-producing enzymes, and as such, the binding pockets for UDP-sugars and DAG remain elusive.

Glycosylglycerolipids play important roles in marine phytoplankton and heterotrophic bacteria under phosphate deprivation. Lipid remodeling reduces the cellular requirement for phosphorus (P), and the glycosylglycerolipids MGlc-DAG/MGlcA-DAG and SQDG replace phospholipids in marine heterotrophic bacteria and marine phytoplankton and cyanobacteria, respectively (16–19). However, little is known about how these glycosylglycerolipids are synthesized. Previous studies have shown that a manganese-dependent metallophosphoesterase, PlcP, is essential for lipid remodeling in marine heterotrophs; that the *plcP* gene is organized in an operon-like structure; and that a putative glycosyltransferase was found downstream of *plcP* in numerous marine heterotrophic bacteria, such as members of the SAR11 clade (17, 20). Our previous work has shown that the GT from the marine bacterium SAR11 ("*Candidatus* Pelagibacter sp." strains HTCC7211 and HTCC1062) is homologous to the Agt GT in *A. tumefaciens* (17).

However, the activity of a SAR11 GT in the synthesis of glycolipids has not been characterized so far.

In this study, we report the detailed biochemical characterization of a glycolipid GT (GT<sub>cp</sub>) from the marine bacterium "*Candidatus Pelagibacter* sp." HTCC7211. Our results showed that GT<sub>cp</sub> has a broad substrate specificity and can synthesize neutral glycolipids (MGlc-DAG and MGal-DAG) and an acidic glycolipid (MGlcA-DAG). GT<sub>cp</sub> represents the first member of the GT4 family of lipid GTs from marine bacteria. In addition, homology modeling and site-directed mutagenesis analyses revealed details of its substrate recognition mechanism and identified key residues involved in the coordination of DAG in a GT4 family glycosyltransferase for lipid biosynthesis.

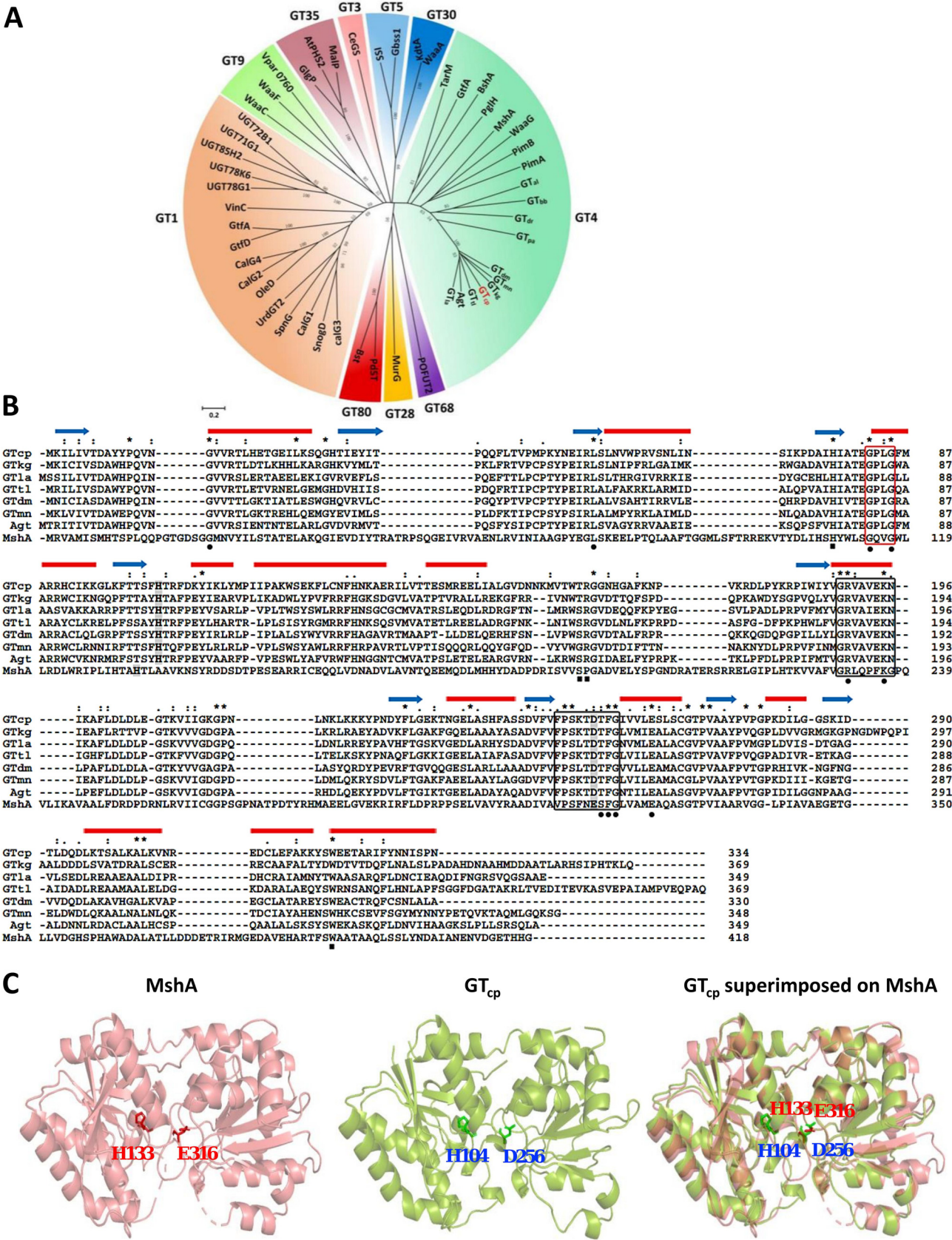
## RESULTS AND DISCUSSION

**GT<sub>cp</sub> and its homologs form a new group in the GT4 family with a GT-B fold structure.** The gene encoding a putative GT<sub>cp</sub> (GenBank accession no. [WP\\_008545403.1](#)) consists of 1,005 bp encoding a peptide of 334 amino acids, containing a GT4-like domain (Fig. 1A). This gene was previously hypothesized to be involved in lipid remodeling in "*Candidatus Pelagibacter* sp." HTCC7211 for the synthesis of MGlc-DAG and MGlcA-DAG (16, 17). Sequence alignment (Fig. 1B) analyses showed that the GT<sub>cp</sub> amino acid sequence has between 46% and 58% sequence identity with the putative GT from *Labrenzia aggregata* (GT<sub>la</sub>; [WP\\_040439323.1](#)), *Thalassospira lucentensis* (GT<sub>tl</sub>; [WP\\_062950653.1](#)), *Methylophaga nitrareducens* (GT<sub>mn</sub>; [WP\\_014706011.1](#)), *Desulfobulbus mediterraneus* (GT<sub>dm</sub>; [WP\\_028584068.1](#)), *Citromicrobium bathyomarinum* JL354 (GT<sub>cb</sub>; [WP\\_010239457.1](#)), and *Kordiimonas gwangyangensis* (GT<sub>kg</sub>; [WP\\_051078133.1](#)) and the characterized Agt (locus tag, atu2297) from *A. tumefaciens* (6). In the phylogenetic analysis, the GT<sub>cp</sub> together with its close homologs (GT<sub>la</sub>, GT<sub>tl</sub>, GT<sub>mn</sub>, GT<sub>dm</sub>, GT<sub>cb</sub>, GT<sub>kg</sub>, and Agt), formed a clade in the GT4 family (Fig. 1A). Sequences from this clade showed low sequence identity (<25%) to other members of the GT4 family, which includes more than 150,000 proteins with at least 22 different enzymatic activities at the time of writing. Purified Agt from *A. tumefaciens* has been found to synthesize MGlc-DAG or MGlcA-DAG with UDP-glucose or UDP-glucuronic acid as the sugar donor, respectively, and the expression of *agt* is known to be induced under phosphate deficiency (17). Neither GT<sub>cp</sub> nor any of its homologs from marine bacteria have been purified or characterized to date.

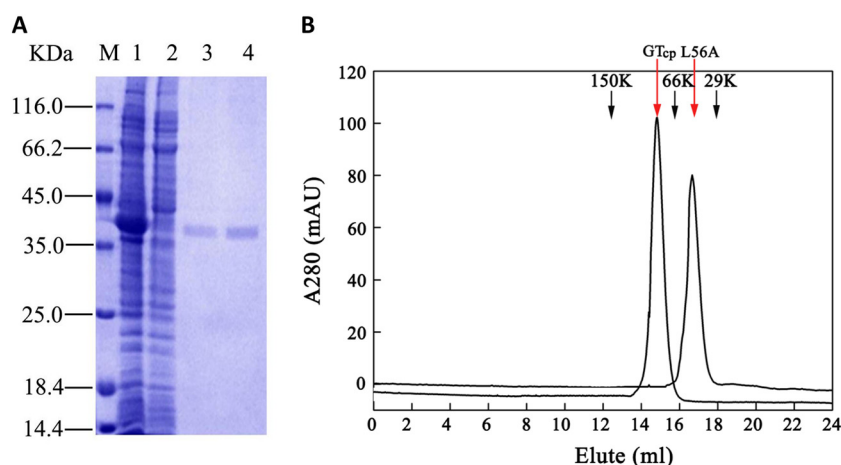
Due to the lack of a three-dimensional structure of DAG-dependent GT4 glycosyltransferases to date, a model of GT<sub>cp</sub> was generated by homology modeling using the X-ray structure of the GT MshA cocrystallized with UDP (Protein Data Bank [PDB] entry [3C4Q](#); 17% sequence identity) from *Corynebacterium glutamicum* as the template (Fig. 1C) (21). MshA, also a member of the GT4 family, catalyzes the first step of the biosynthesis of mycothiol in actinobacteria using UDP-N-acetylglucosamine (UDP-GlcNAc) as the sugar donor (21). Using the PDBFold server, we compared the crystal structure of MshA with the modeled structure of GT<sub>cp</sub>. The predicted structure of GT<sub>cp</sub> includes a GT-B fold consisting of two Rossmann-like  $\beta$ - $\alpha$ - $\beta$  domains, with the N-terminal domain (residues 1 to 160 and residues 320 to 332) and C-terminal domain (residues 169 to 315), separated by a large cleft that includes the catalytic center. The same GT-B fold is also found in several members of the GT4 family enzymes, including MshA and PimA (Fig. 1A) (21, 22). The substrate binding site of the sugar donor is located mainly in the C-terminal domain, where the sugar donor forms a number of hydrogen bonds with the protein. Given that the two enzymes (i.e., MshA and GT<sub>cp</sub>) share only 17% overall sequence identity, the alignment was manually corrected by incorporating information such as predicted secondary structures and conserved functional residues. Multiple sequence alignment of GT<sub>cp</sub> and its homologs revealed that GT<sub>cp</sub> contains a catalytic dyad composed of His104-Asp256, two conserved UDP-sugar binding motifs (GRVAXEKN and FPSXTDTFG), and a conserved Gly-rich motif, of which all are commonly found in the GT4 family (Fig. 1B) (23).

**Cloning of putative gene encoding GT<sub>cp</sub> and functional verification.** The recombinant plasmid pET22b-GT<sub>cp</sub> was constructed to purify the enzyme for determination of





**FIG 1** Multiple sequence alignment and functional domain analyses of GT<sub>cp</sub> protein. (A) A phylogenetic tree of GT<sub>cp</sub> and its homologs with known 3-dimensional X-ray structures of GTs. Sequences and structures of GTs were obtained from the NCBI database and the PDB database, including GT1 (calG3, (Continued on next page)



**FIG 2** (A) Overexpression and purification of  $GT_{cp}$  proteins from “*Candidatus Pelagibacter* sp.” HTCC7211. M, protein molecular weight marker. Lane 1, cell-free supernatant induced with isopropyl  $\beta$ -D-1-thiogalactopyranoside (IPTG); lane 2, cell-free supernatant without IPTG induction; and lanes 3 to 4, purified  $GT_{cp}$  (molecular weight estimated to be ~38 kDa). (B) Gel filtration analysis of the wild-type  $GT_{cp}$  and the mutant L56A. The red arrows indicate the eluted position of  $GT_{cp}$  and the L56A mutant, and the black arrows indicate protein markers (from left to right: alcohol dehydrogenase [150 kDa, 12.35 ml], albumin [66 kDa, 15.83 ml], and carbonic anhydrase [29 kDa, 17.95 ml]).

its catalytic properties. Soluble expression of His-tagged  $GT_{cp}$  was achieved in *Escherichia coli* BL21(DE3) by adding 0.5 mM isopropyl- $\beta$ -D-thiogalactopyranoside (IPTG). The recombinant  $GT_{cp}$  was purified to homogeneity by Ni-nitrilotriacetic acid (NTA) affinity followed by gel filtration chromatography using the Superdex 200 column. In SDS-PAGE analyses, the purified recombinant His-tagged  $GT_{cp}$  was visible as a major band with a calculated mass of 38 kDa (Fig. 2A). The process used to purify  $GT_{cp}$  is summarized in Table 1. The enzyme was purified 3.5-fold with a yield of 17.1% and a specific activity of 47.1 U/mg. The predicted molecular mass of the monomeric form of  $GT_{cp}$  is 38 kDa, but  $GT_{cp}$  eluted as a peak corresponding to a molecular mass of approximately 70 kDa in the gel filtration chromatography experiments (Fig. 2B). These results suggested that  $GT_{cp}$  forms a dimer in solution, consistent with the proposed mechanism that oligomerization is a major factor contributing to the biochemical function and enzymatic activity of GTs (23).

We used two methods to determine the activity of purified  $GT_{cp}$ . Previous studies have shown that thin-layer chromatography (TLC) can resolve these selected glycosylcerolipid standards (14, 15). Indeed, as shown in Fig. 3A, the products of the enzymatic reaction with MGlc-DAG, MGal-DAG, and MGlcA-DAG were observed by staining with sulfuric acid-methanol-water and corresponded to the standard markers. These results showed that  $GT_{cp}$  is able to transfer galactose, glucose, and hexuronic acid to the DAG acceptor using the respective UDP-sugars.

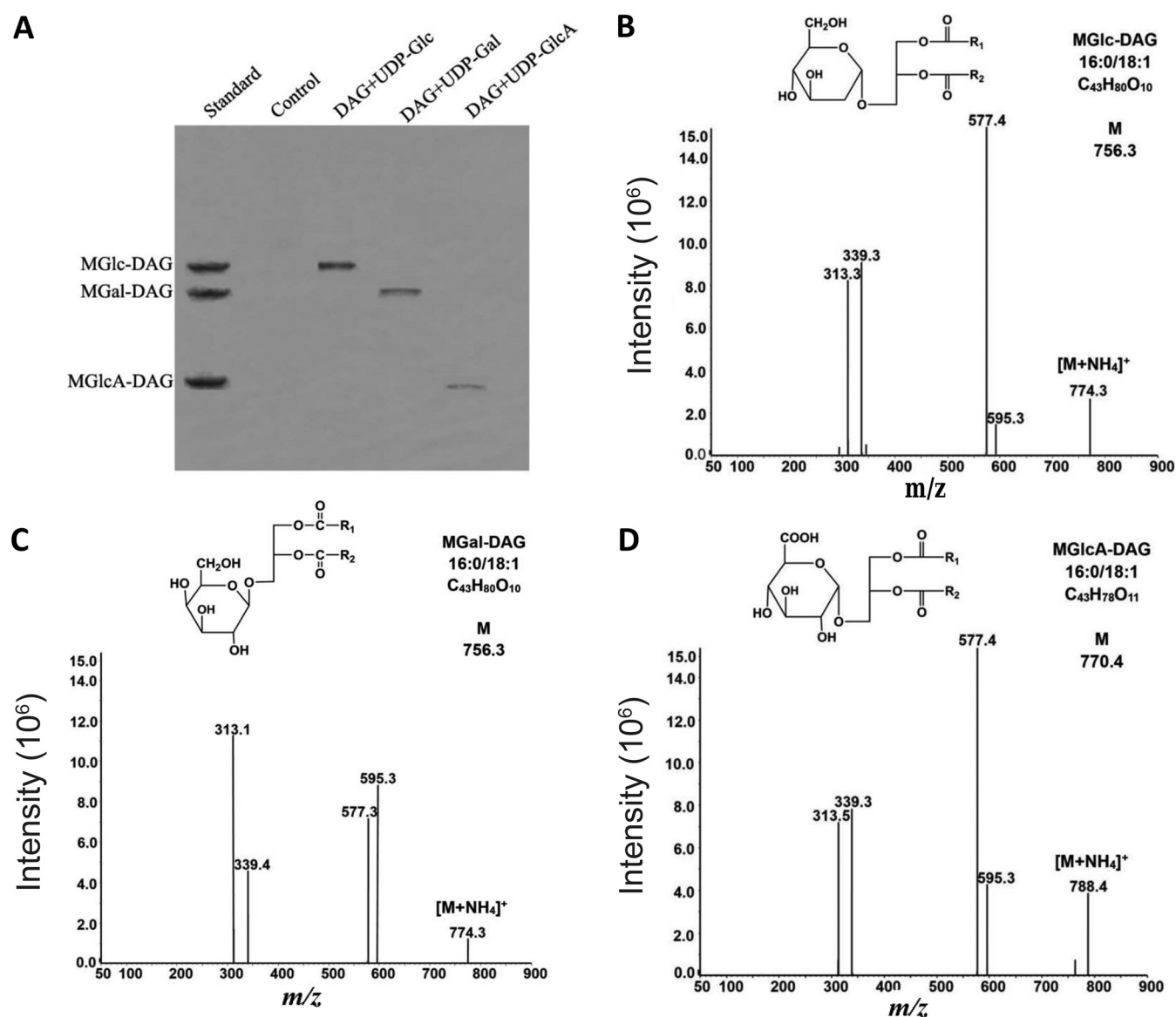
#### FIG 1 Legend (Continued)

SnogD, CalG1, SpnG, UrdGT2, OleD, CalG2, CalG4, GtfD, GtfA, Vinc, UGT78G1, UGT78K6, UGT85H2, UGT71G1, and UGT72B1), GT3 (CeGs), GT5 (Gbss1 and SSI), GT9 (WaaC, WaaF, and Vpar 0760), GT28 (MurG), GT30 (WaaA and KdtA), GT35 (AtPHS2, MalP, and GlgP), GT68 (POFUT2), GT80 (Pdst and Bst), GT4 (PimA, PimB, WsaF, GtfA, TarM, PglH, MshA, BshA, and WaaG),  $GT_{cp}$  and its homologs ( $GT_{lar}$ ,  $GT_{tir}$ ,  $GT_{mnf}$ ,  $GT_{dmr}$ ,  $GT_{cbr}$ ,  $GT_{kgr}$ ,  $GT_{alr}$ ,  $GT_{bbr}$ ,  $GT_{drf}$ ,  $GT_{par}$ , and Agt).  $GT_{pa}$  is the GT4 glycosyltransferase of *Pseudomonas* sp. PA14 (Fig. S1). (B) Multiple sequence alignment for  $GT_{cp}$  and its homologs using Clustal W program with manual adjusting, including  $GT_{cp}$  from “*Candidatus Pelagibacter* sp.” HTCC7211 (WP\_008545403.1),  $GT_{la}$  from *Labrenzia aggregata* (WP\_040439323.1),  $GT_{ti}$  from *Thalassospira lucentensis* (WP\_062950653.1),  $GT_{mn}$  from *Methylophaga nitratireducens* (WP\_014706011.1),  $GT_{dm}$  from *Desulfobulbus mediterraneus* (WP\_028584068.1),  $GT_{cb}$  from *Citromicrobium bathyomarinum* JL354 (WP\_010239457.1),  $GT_{kg}$  from *Kordiimonas gwangyangensis* (WP\_051078133.1), the characterized Agt from *A. tumefaciens* (atu2297), and MshA from *C. glutamicum* (WP\_143854623.1). Red bars represent  $\alpha$ -helical regions, and blue arrows represent  $\beta$ -sheet from MshA (Protein Data Bank [PDB] entry 3C4Q). Residues interacting with UDP-sugar and DAG are indicated by the closed black circles and squares, respectively. Catalytic dyad composed of His104-Asp256 is shaded in gray. Black boxed regions and red boxed regions in the alignment indicated two conserved UDP-sugar binding motifs of GRVAXEKN and FPSXTDTFG and a conserved Gly-rich motif, respectively. (C) Homology modeling showing the predicted structure of  $GT_{cp}$  and the catalytic dyad composed of His104-Asp256. The signature glutamate residue in MshA (Glu316) is substituted to an aspartate residue (Asp256) in  $GT_{cp}$ .

**TABLE 1** Purification of the recombinant GT<sub>cp</sub> from "*Candidatus Pelagibacter sp.*" HTCC7211

Step	Total protein (mg)	Total activity (U)	Specific activity (U/mg)	Purification fold	Yield (%)
Crude cell extract	34.8	469	13.4	1	100
Ni-NTA affinity	1.9	85	44.7	3.3	18.1
Superdex 200 gel filtration	1.7	80	47.1	3.5	17.1

For structural identification, the glycolipids were analyzed by liquid chromatography-mass spectrometry (LC-MS). LC-MS analyses (in the positive ion mode) detected an ammonium adduct (Fig. 3B to D), and fragmentation spectra for monohexuronosyl DAGs (MGlc-DAG and MGal-DAG) and MGlcA-DAG were obtained from the products of the GT<sub>cp</sub>-catalyzed reactions. The calculated  $m/z$  of the parental ion of MGlc-DAG, MGal-DAG, and MGlcA-DAG was 756.3, 756.3, and 770.4, respectively. The two species differed in the neutral loss corresponding to the polar head group (179.0 and 193.1  $m/z$  for the loss of the hexosyl group and the hexuronic acid group,



**FIG 3** Functional characterization of recombinant GT<sub>cp</sub>. (A) TLC of the enzymatic reaction products with different UDP-sugar donors and diacylglycerol (DAG) as the acceptor by staining with sulfuric acid-methanol-water (45:45:10). (B to D) LC-MS of fragmentation spectra for monohexuronosyl DAGs (MGlc-DAG and MGal-DAG), and MGlcA-DAG were obtained from the products of the GT<sub>cp</sub>-catalyzed reaction.

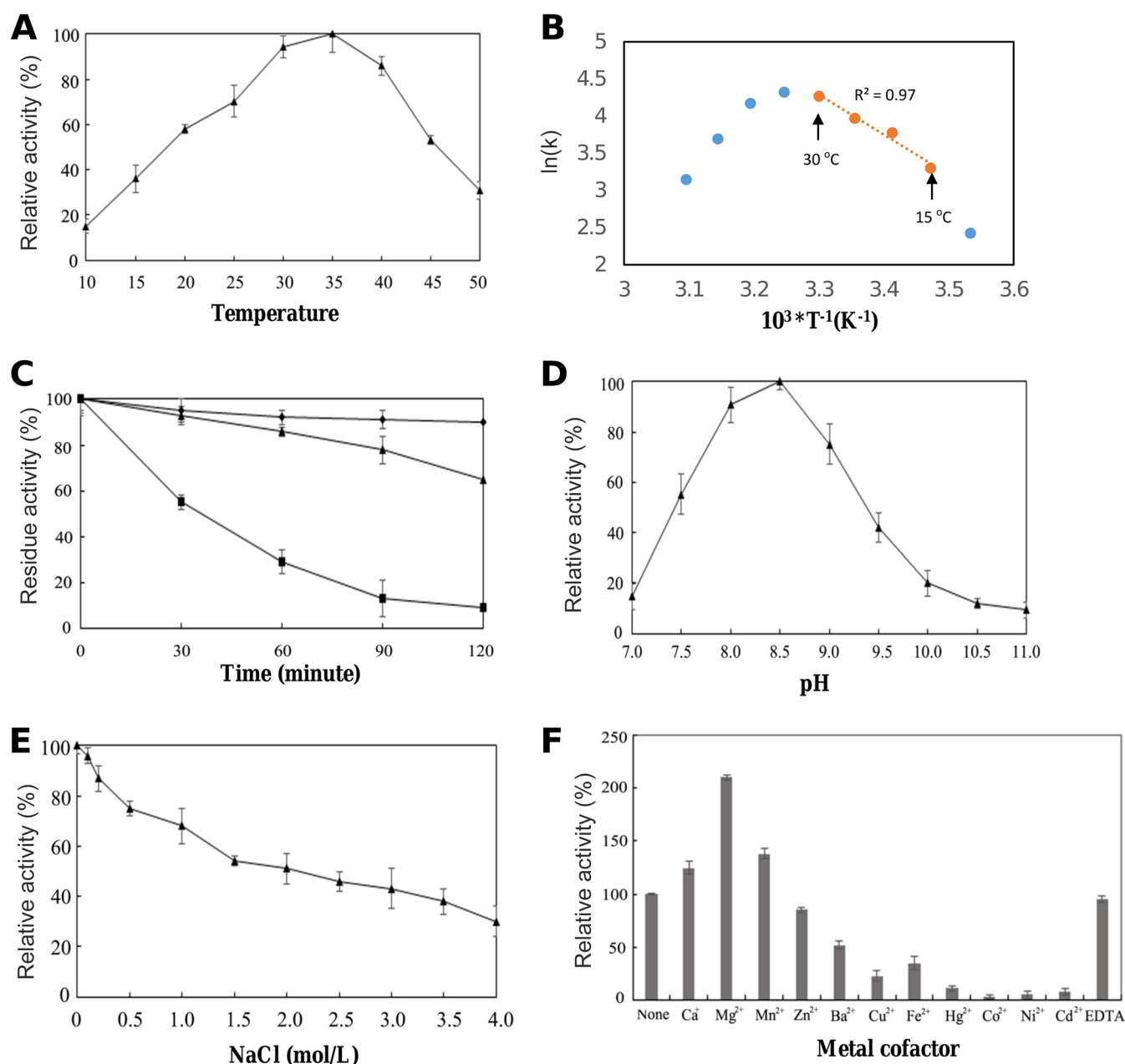


respectively). In each case, this loss yielded a DAG-16:0/18:1 ( $m/z$ , 577.3 or 577.4). A further two peaks corresponded to monoacylglycerol with glyceryl-16:0 ( $m/z$ , 313.3 or 313.5) and -18:1 ( $m/z$ , 339.3 or 339.4) fatty acids, respectively. These results demonstrated that GT<sub>cp</sub> shows high enzymatic activity toward the synthesis of MGlc-DAG, MGal-DAG, and MGlcA-DAG from DAG and UDP-sugars. Several bacterial lipid GTs from *M. loti*, *A. tumefaciens*, *Mycoplasma pneumonia*, and *Mycoplasma genitalium* have been found to synthesize different glycolipids (DGal-DAG, GlcGal-DAG, and triglycosyl DAGs) using UDP-Glu and UDP-galactose (UDP-Gal) as sugar donors (14, 15, 24, 25). The *A. tumefaciens* GT Agt, which synthesizes neutral glycolipid (MGlc-DAG) and acidic glycolipid (MGlcA-DAG) with UDP-Glc or UDP-GlcA as the sugar donor, respectively, has been isolated and characterized (17). To the best of our knowledge, GT<sub>cp</sub> is the first bacterial lipid GT acting as a multifunctional enzyme to synthesize MGlc-DAG, MGal-DAG, and MGlcA-DAG with different UDP-sugars as donors. It remains to be seen whether bifunctional/multifunctional GT4 enzymes involved in glycolipid synthesis are a common trait in this group. At least one member of this family, the GT4 homologue in *Pseudomonas* sp., appears to be specific for UDP-glucose and does not accept UDP-galactose nor UDP-glucuronic acid as the substrate (see Fig. S1 in the supplemental material).

**Catalytic properties of GT<sub>cp</sub>.** The catalytic activity of GT<sub>cp</sub> was tested with UDP-Glc as the sugar donor substrate and DAG as the potential acceptor substrate. The effect of temperature on GT<sub>cp</sub> activity was determined in the range of 10 to 50°C (Fig. 4A). GT<sub>cp</sub> showed maximum activity at around 35°C and more than half of maximum activity at 20 to 45°C. Calculation of the activation energy ( $E_a$ ) using the Arrhenius plot in a semilogarithmic form [ $\ln(k) \rightarrow 1/T$ ] was shown in Fig. 4B, and the slope of the plot was used to calculate the activation energy which was  $E_a = 44.9 \text{ kJ mol}^{-1}$ . The thermostability of GT<sub>cp</sub> was evaluated at three different temperatures (30°C, 40°C, and 50°C) with increasing incubation times up to 120 min. Most of the enzyme activity was maintained after incubation at 40°C for at least 120 min, whereas incubation at 50°C for 30 min reduced activity by approximately 50% (Fig. 4C). To investigate the effect of pH on the enzymatic activity of GT<sub>cp</sub>, the enzymatic reaction was evaluated in different buffers (pH 7.0 to 11.0). The maximum activity of GT<sub>cp</sub> was at pH 8.5, and it retained more than 50% of maximum activity between pH 7.5 and 9.0 (Fig. 4D). Considering the significance of NaCl for marine enzymes, enzyme activity was determined in the presence of NaCl at different concentrations. The recombinant GT<sub>cp</sub> maintained 54% of its maximum activity in the presence of 1.5 M NaCl and 30% of its maximum activity when the NaCl concentration was increased to 4 M (Fig. 4E).

To investigate the substrate specificity of GT<sub>cp</sub> for DAGs, different species of various chain length of DAGs were tested (Table 2). The  $K_m$  and  $k_{cat}$  values of GT<sub>cp</sub> were calculated from Hanes-Woolf plots and the Michaelis-Menten equation. For the DAGs with saturated fatty acid chains (di8:0, di10:0, di12:0, di14:0, di16:0, and di18:0), the  $K_m$ ,  $k_{cat}$ , and  $k_{cat}/K_m$  values increased with increasing acyl chain length. GT<sub>cp</sub> showed the higher activities for the unsaturated DAGs than for the saturated DAGs. The most preferred DAG substrate for GT<sub>cp</sub> was C<sub>16:0</sub>/C<sub>18:1</sub> DAG, consistent with the fact that C<sub>16:0</sub> and C<sub>18:1</sub> fatty acids are common in marine bacteria (16, 17).

The Michaelis-Menten kinetic parameters for GT<sub>cp</sub> were determined using UDP-Glc, UDP-Gal, and UDP-GlcA as the sugar donors (Table 3). The  $K_m$  value for UDP-Glc (82  $\mu\text{M}$ ) was higher than those for UDP-Gal and UDP-GlcA, consistent with the fact that UDP-Glc is the preferred substrate under physiological conditions (26). The  $k_{cat}/K_m$  value for GT<sub>cp</sub> toward different sugar donors followed the order UDP-Glc ( $71.4 \pm 2.7$ ) > UDP-GlcA ( $59.6 \pm 3.6$ ) > UDP-Gal ( $32.2 \pm 2.9$ ). UDP-xylose, UDP-rhamnose, UDP-mannose, and UDP-fructose were also tested with C<sub>16:0</sub>/C<sub>18:1</sub> DAG, which showed no activity. Thus, GT<sub>cp</sub> exhibited the highest enzymatic activity for UDP-Glc among the sugar donors tested. In contrast with GT<sub>cp</sub>, the Pgts from *M. loti* and *A. tumefaciens* favor uridine UDP-Gal over UDP-Glc (14, 15). A comparison of GT<sub>cp</sub> kinetics with other biochemically characterized GT4 glycosyltransferases is listed in Table 4 (21, 27–32).



**FIG 4** Biochemical characterization of  $GT_{cp}$ . (A) Effect of temperature on the activities of  $GT_{cp}$ . (B) Arrhenius plot of  $GT_{cp}$ . The activation energy of the reaction  $E_a = 44.9 \text{ kJ mol}^{-1}$  could be determined from the slope of the regression curve with data from 15°C to 30°C ( $R^2 = 0.97$ ). At higher temperatures, enzyme activities decreased, likely due to denaturation of the enzyme. (C) Thermostability of  $GT_{cp}$ . The residual enzyme activity was measured after incubation of the purified enzyme at 30°C (diamonds), 40°C (triangles), and 50°C (boxes), respectively. (D) Effect of pH on the activities of  $GT_{cp}$ . (E) Effect of NaCl on the activities of  $GT_{cp}$ . The enzyme was incubated in buffers containing different concentrations of NaCl (0 to 4 M) at 4°C for 1 h. Residual activity was measured under optimal conditions. (F) Effect of metal ions (5 mM) on the activities of  $GT_{cp}$ . The values are means of three independent experiments.

**Metal ions improve enzyme activity of  $GT_{cp}$ .** The effect of various metal ions on the enzyme activity of  $GT_{cp}$  is shown in Fig. 4F. Among the tested metal ions (5 mM),  $Mg^{2+}$ ,  $Ca^{2+}$ , and  $Mn^{2+}$  significantly stimulated  $GT_{cp}$  activity by up to 223%, 125%, and 138%, respectively, although the as-isolated enzyme is already active. Furthermore, EDTA did not significantly affect the enzymatic activity of  $GT_{cp}$  after incubation for 60 min at room temperature. The activity of  $GT_{cp}$  was decreased by  $Ba^{2+}$ ,  $Cu^{2+}$ , and  $Fe^{2+}$  to 52%, 23%, and 35%, respectively. Moreover,  $Hg^{2+}$ ,  $Cd^{2+}$ ,  $Ni^{2+}$ , and  $Co^{2+}$  completely abolished enzyme activity. This may have resulted from the binding of these



**TABLE 2** Kinetic parameters of GT<sub>cp</sub> using UDP-Glc, UDP-Gal, and UDP-GlcA as sugar donors and different DAGs as acceptors<sup>a</sup>

Substrate	$k_{cat}$ (min <sup>-1</sup> )	$K_m$ (μM)	$k_{cat}/K_m$ (min <sup>-1</sup> mM <sup>-1</sup> )
UDP-Glc/DAG (C <sub>16:0</sub> /C <sub>18:1</sub> )	5.9 ± 0.4	82.0 ± 0.3	71.9 ± 2.7
UDP-Gal/DAG (C <sub>16:0</sub> /C <sub>18:1</sub> )	1.0 ± 0.3	32.0 ± 0.4	31.3 ± 2.9
UDP-GlcA/DAG (C <sub>16:0</sub> /C <sub>18:1</sub> )	4.1 ± 0.1	68.0 ± 0.5	60.3 ± 3.6
UDP-Glc/DAG (C <sub>18:0</sub> /C <sub>20:4</sub> )	4.8 ± 0.2	77.1 ± 0.4	62.2 ± 0.9
UDP-Glc/DAG (C <sub>18:0</sub> /C <sub>18:2</sub> )	6.2 ± 1.5	102.5 ± 0.8	60.5 ± 1.8
UDP-Glc/DAG (di18:1)	5.0 ± 0.8	90.5 ± 1.6	55.2 ± 2.2
UDP-Glc/DAG (di18:0)	3.6 ± 0.7	75.3 ± 1.2	47.8 ± 1.3
UDP-Glc/DAG (di16:0)	1.7 ± 1.4	55.2 ± 0.9	30.8 ± 2.5
UDP-Glc/DAG (di14:0)	1.0 ± 2.2	45.9 ± 1.1	21.7 ± 2.8
UDP-Glc/DAG (di12:0)	0.5 ± 0.4	21.6 ± 0.8	13.9 ± 1.7
UDP-Glc/DAG (di10:0)	ND	ND	ND
UDP-Glc/DAG (di8:0)	ND	ND	ND
UDP-Glc/DAG (C <sub>16:0</sub> /C <sub>18:1</sub> ) <sup>b</sup>	8.3 ± 0.3	58.0 ± 0.6	143.1 ± 3.9

<sup>a</sup>The values are means of three independent experiments ± standard deviations. ND, not detectable.<sup>b</sup>The kinetic parameters were determined using purified GT<sub>cp</sub> in the presence of Mg<sup>2+</sup> (5 mM).

metal ions to the -SH, -CO, and -NH moieties of the amino acids of GT<sub>cp</sub>, leading to structural changes and inactivation (33).

Given that Mg<sup>2+</sup> markedly improved the activity of GT<sub>cp</sub>, we determined the metal content of the purified enzyme by inductively coupled plasma-mass spectrometry (ICP-MS) and found that the Mg<sup>2+</sup>: protein molar ratio was <0.03 (see Table S1 in the supplemental material). Similarly, no substantial amount of Ca, Mn, or Zn was found in GT<sub>cp</sub>, suggesting that this enzyme is unlikely a metalloprotein. The kinetic parameters were subsequently determined using purified GT<sub>cp</sub> in the presence and absence of Mg<sup>2+</sup> (Table 3). In the presence of excess DAG, the  $K_m$  value of GT<sub>cp</sub> decreased from 82 μM to 58 μM with the addition of Mg<sup>2+</sup>, indicating that GT<sub>cp</sub> has a higher affinity for UDP-Glc in the presence of Mg<sup>2+</sup>. The catalytic efficiency was nearly 2.0-fold higher than that in the absence of Mg<sup>2+</sup>. In the presence of excess UDP-Glc, the  $K_m$  was almost unchanged with/without Mg<sup>2+</sup>. Metal ions are important regulators of physiological functions and contribute to the preservation of the structural integrity of some proteins (34). In contrast to GT-A fold GTs, GT-B fold GTs, including GT<sub>cp</sub>, are metal ion independent (23). However, some studies have found that metal ions also change GT-B fold activity, such as GGT58A1 from *Absidia coerulea* (35), UGT59A1 from *Rhizopus japonicus*, Bs-PUGT from *Bacillus subtilis* P118 (36), and human POFUT2 (37). In some cases, the metal ion simultaneously interacts with both the enzyme and the sugar donor in the active site, which causes the glucosyl donor to realign in the active site, and thus affects the activity of the enzyme (38). Interestingly, although human POFUT2 is not a metalloprotein, Mg<sup>2+</sup>, Mn<sup>2+</sup>, and Ca<sup>2+</sup> are also known to enhance the enzyme activity by facilitating the release of product from the enzyme (37). Therefore, it is tempting to speculate that the addition of Mg<sup>2+</sup> may also help enhance GT<sub>cp</sub> catalysis in a similar manner.

**Insights into the structure and glycosylation mechanism of GT<sub>cp</sub>.** GT<sub>cp</sub> along with the previously characterized Agt form a group in the GT4 family and are GT-B fold GTs (Fig. 1, Fig. 5A). The GT-B fold GTs are thought to employ the so-called retaining glycosylation for adding the sugar unit to the substrate, although some researchers

**TABLE 3** Kinetic parameters of GT<sub>cp</sub> using UDP-Glc, UDP-Gal, and UDP-GlcA as sugar donors and C<sub>16:0</sub>/C<sub>18:1</sub> DAG as acceptors<sup>a</sup>

Substrate	$k_{cat}$ (min <sup>-1</sup> )	$K_m$ (μM)	$k_{cat}/K_m$ (min <sup>-1</sup> mM <sup>-1</sup> )
UDP-Glc	5.85 ± 0.4	82 ± 0.3	71.4 ± 2.7
UDP-Gal	1.03 ± 0.3	32 ± 0.4	32.2 ± 2.9
UDP-GlcA	4.05 ± 0.1	68 ± 0.5	59.6 ± 3.6
UDP-Glc (with 5 mM Mg <sup>2+</sup> )	8.25 ± 0.3	58 ± 0.6	142.2 ± 3.9

<sup>a</sup>The values are means ± standard deviations of three independent experiments.

**TABLE 4** Kinetic parameters of GT<sub>cp</sub> in comparison with GT4 glycosyltransferases using a different sugar donor<sup>a</sup>

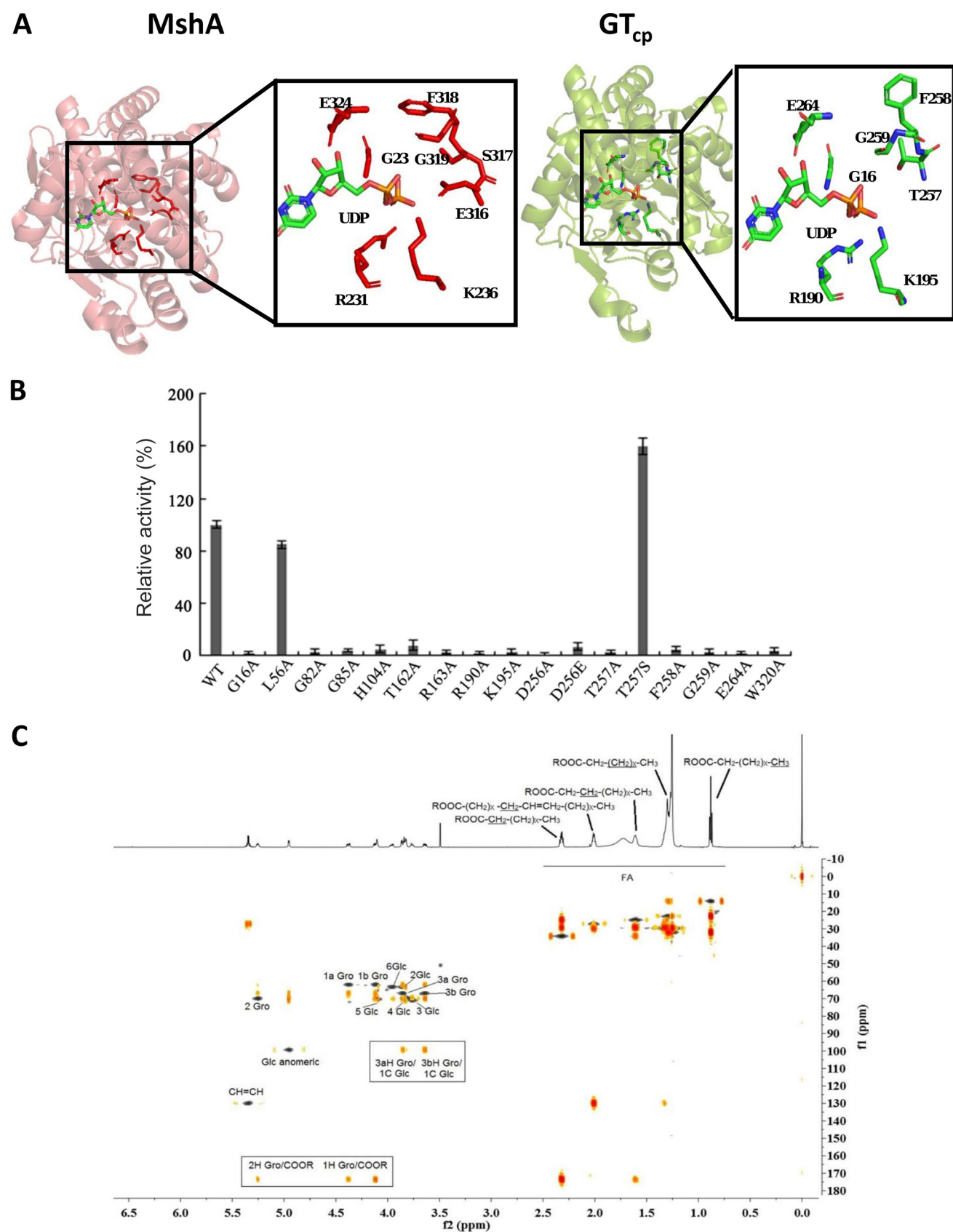
Parameter	Kinetics of GT type with donor substrate							
	UDP-Glc	GDP-Man		UDP-GlcNAc				
	GT <sub>cp</sub>	PimA	PimB	MshA	TarM	BshA	GtfA	PglH
$K_m$ ( $\mu$ M)	82 $\pm$ 0.3	18 $\pm$ 2	19.0 $\pm$ 4.6	0.208 $\pm$ 0.017	65 $\pm$ 10	180 $\pm$ 50	11.8 $\pm$ 1.5	2.6 $\pm$ 0.3
$k_{cat}$ ( $\text{min}^{-1}$ )	5.8 $\pm$ 0.4				126 $\pm$ 10	78.6	7.35 $\pm$ 0.42	4.4 $\pm$ 0.2
$k_{cat}/K_m$ ( $\text{min}^{-1} \text{mM}^{-1}$ )	71.4 $\pm$ 2.7					6.8	0.63	1.7 $\pm$ 0.2

<sup>a</sup>The values are means  $\pm$  standard deviations of three independent experiments. UDP-GlcNAc, UDP-N-acetylglucosamine; GDP-Man, GDP-mannose.

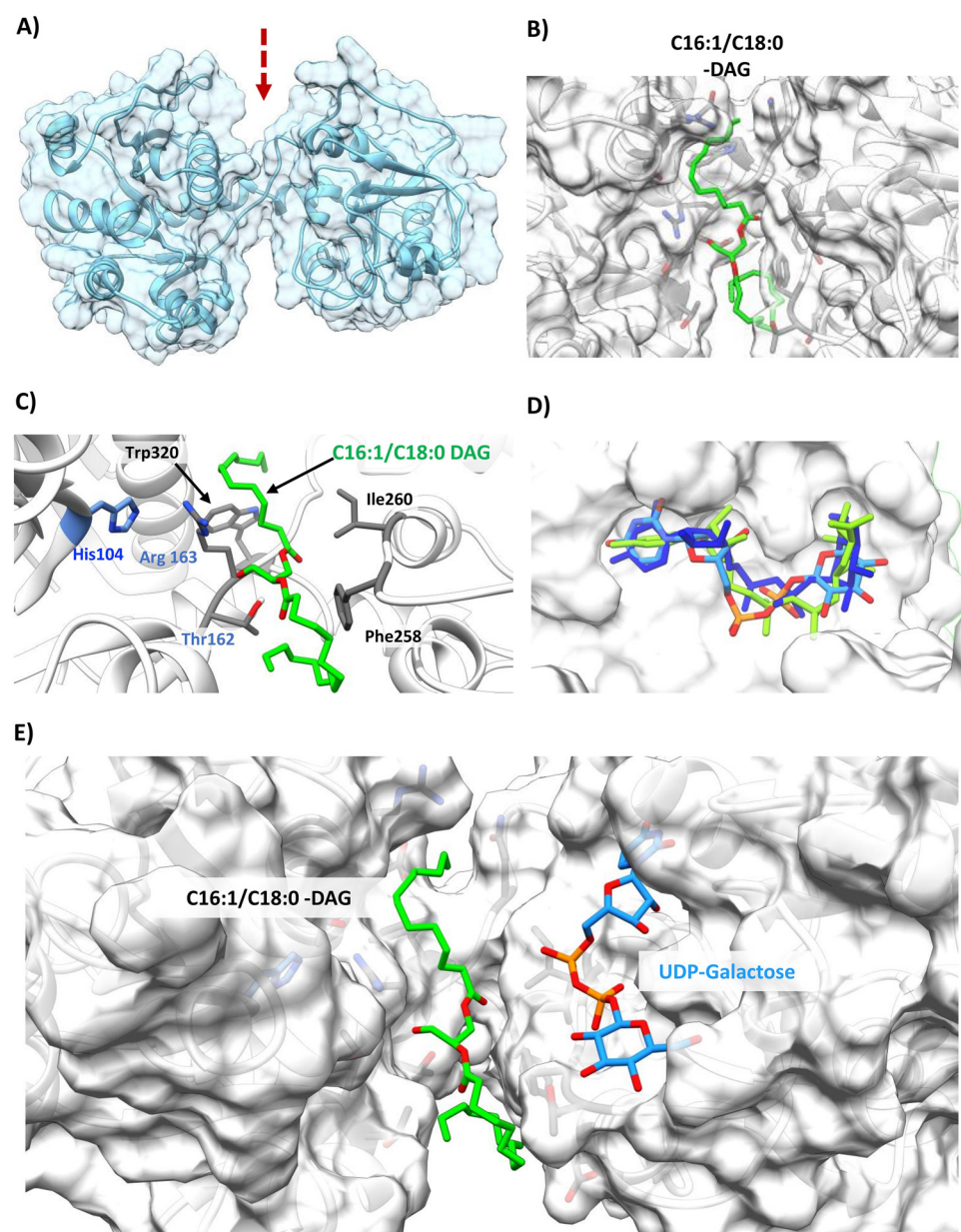
have proposed an alternative internal return ( $S_Ni$ -like) mechanism (23, 39). In the latter model, the nucleophilic attack and the departure of the leaving group occur on the same face of the sugar and involve the formation of a short-lived oxocarbenium-like transition state with asynchronous acceptor glycoside bond formation and phosphate bond breakdown (40–42). According to the latter model, two conserved amino acid residues are important for catalysis (e.g., Glu316 and His109 in MshA). In MshA, Glu316 and His109 act as catalytic nucleophiles in the  $S_Ni$ -like mechanism, and form hydrogen bonds with the OH-3 and OH-6 of the glucosyl moiety, respectively (21). Indeed, we also found the corresponding residues (Asp256 and His104) in GT<sub>cp</sub> (Fig. 1B). Asp256 and His104 fulfil a critical role in GT<sub>cp</sub> catalysis because their substitution to Ala completely abolished catalytic activity (Fig. 5B). Interestingly, mutation of the conserved Asp residue at position 256 to Glu (D256E) reduced activity by more than 90% even though both Asp and Glu residues have a carboxylic acid moiety. Asp256 is essential for the enzymatic activity of GT<sub>cp</sub> and cannot be replaced by Glu, indicating that the length of the side chain of this residue is important for the activity of GT<sub>cp</sub>. These results suggested that Asp256 and His104 play important functions in glycosyl transfer while maintaining the  $\alpha$ -configuration of the anomeric carbon. To confirm the glycosylated position and the anomeric stereoselectivity, MGlc-DAG was purified and analyzed by <sup>1</sup>H-nuclear magnetic resonance (NMR) (Fig. 5C). The glucosyl moiety of the glycolipid was suggested by the characteristic signals for Glc-H1 ( $\delta$  H 5.17, 1H) and Glc-H2-6 ( $\delta$  H 3.20 to 3.70, 6 $\times$ H) by <sup>1</sup>H NMR. The location of the glucosyl moiety was indicated by the correlation between Glc-H1 and C-3 ( $\delta$  H 136.0) in the heteronuclear multiple-bond correlation (HMBC) spectrum. The large coupling constant ( $J$  = 3.7 Hz) of Glc-H1 revealed the  $\alpha$ -D-configuration of the glucosidic linkage.

In terms of the donor binding pocket, GT4 family GTs have two highly conserved sequence motifs in their C-terminal domain that are involved in the binding of the sugar nucleotide with the UDP moiety (22, 28, 43). The sequence alignment analyses indicated that GT<sub>cp</sub> and its homologs have these two conserved motifs in the donor binding pocket (Fig. 1B), consistent with other GT4 proteins. The highly conserved sites were further investigated to illuminate the mechanism of sugar donor binding (Fig. 5A). Each of the potential active sites in the donor binding pocket (Gly16, Gly82, Gly85, Arg190, Lys195, Thr257, Phe258, Gly259, and Glu264) was mutated to Ala. Enzymatic activity of GT<sub>cp</sub> in all the mutants was almost completely abolished, supporting our assumption from the homology model that these residues are integral to the sugar donor binding site (Fig. 5B). Thr257 in the structure of GT<sub>cp</sub> corresponds to MshA residue Ser317, which interacts with the 4-OH of the sugar moiety and forms hydrogen bonds (21). When the residue at position 257 (Thr) in GT<sub>cp</sub> was substituted with Ser (T257S), the T257S mutant exhibited approximately 1.6-times higher activity than that of wild-type GT<sub>cp</sub>. Compared with the parental residue Thr, Ser has one fewer methyl group; therefore, there is more free space between the Ser residue and the substrate (44). The generated space might allow more room for the appropriate interaction between the Ser residue and the sugar donor. These findings indicate that the mechanism of the sugar donor recognition of GT<sub>cp</sub> is similar to that of known GT-B fold GTs, such as MshA. That is, the conserved key residues would form hydrogen bonds with the UDP part and interact with the sugar moiety of the sugar donor.

In order to identify potential residues involved in binding of DAG, we docked C<sub>16:1</sub>/



**FIG 5** Homology modeling prediction of the UDP-sugar donor binding pocket in the GT<sub>cp</sub>. (A) UDP-sugar binding site of MshA (21) and predicted UDP-sugar binding site of GT<sub>cp</sub>. (B) Mutational analysis of the key amino acids involved in the catalytic glycosylation reactions of GT<sub>cp</sub>. (C) Overlay of <sup>1</sup>H, heteronuclear single quantum correlation (HSQC) and heteronuclear multiple-bond correlation (HMBC) spectra of MGLc-DAG.



**FIG 6** Identification of key residues for diacylglycerol (DAG) binding in the GT<sub>cp</sub>. (A) Homology modeling prediction of the open model based on template 2R60. The arrow indicates the wide cleft. (B) C<sub>16:1</sub>/C<sub>18:0</sub>-DAG (green) docked in GT<sub>cp</sub> in the groove (shown as a transparent surface). (C) A detailed depiction of key coordinating hydrophobic (black) and polar (light blue) residues for DAG. His104, Thr162, Arg163, and Trp320 are crucial for enzyme activity (Fig. 5B). (D) An overlay of UDP-glucose (blue), UDP-galactose (cyan), and UDP-glucuronic acid (green) showing all three ligands can occupy the same binding site in similar poses in the DAG-docked GT<sub>cp</sub>. (E) A top-down view of DAG and UDP-galactose shown docked parallel to each other in their respective binding pockets/grooves.

C<sub>18:0</sub> DAG into the homology model of GT<sub>cp</sub>. DAG is predicted to lay in the groove of the open structure in the model (Fig. 6A and B). Docked DAG interacts with several key residues primarily through hydrophobic interactions but also some polar interactions, e.g., His104, Thr162, Arg163, and Trp320 (Fig. 6C), of which His104, Arg163, and Trp320 are strictly conserved in GT<sub>cp</sub> in a range of bacteria (Fig. 1B). Indeed, the His104, Arg163, and Trp320 mutants are inactive, supporting a key role in GT<sub>cp</sub> catalysis (Fig. 5B). Subsequent independent docking of the three UDP-sugars predicted the identical binding pocket for all three (Fig. 6D) in the presence of already docked DAG, and with



UDP-galactose as an example, we observe the DAG and UDP-sugars positioned parallel to one another in the “open state” (Fig. 6E). To the best of our knowledge, our study provides the first insight of the binding pocket of DAG in a GT4 glycosyltransferase involved in glycolipid biosynthesis.

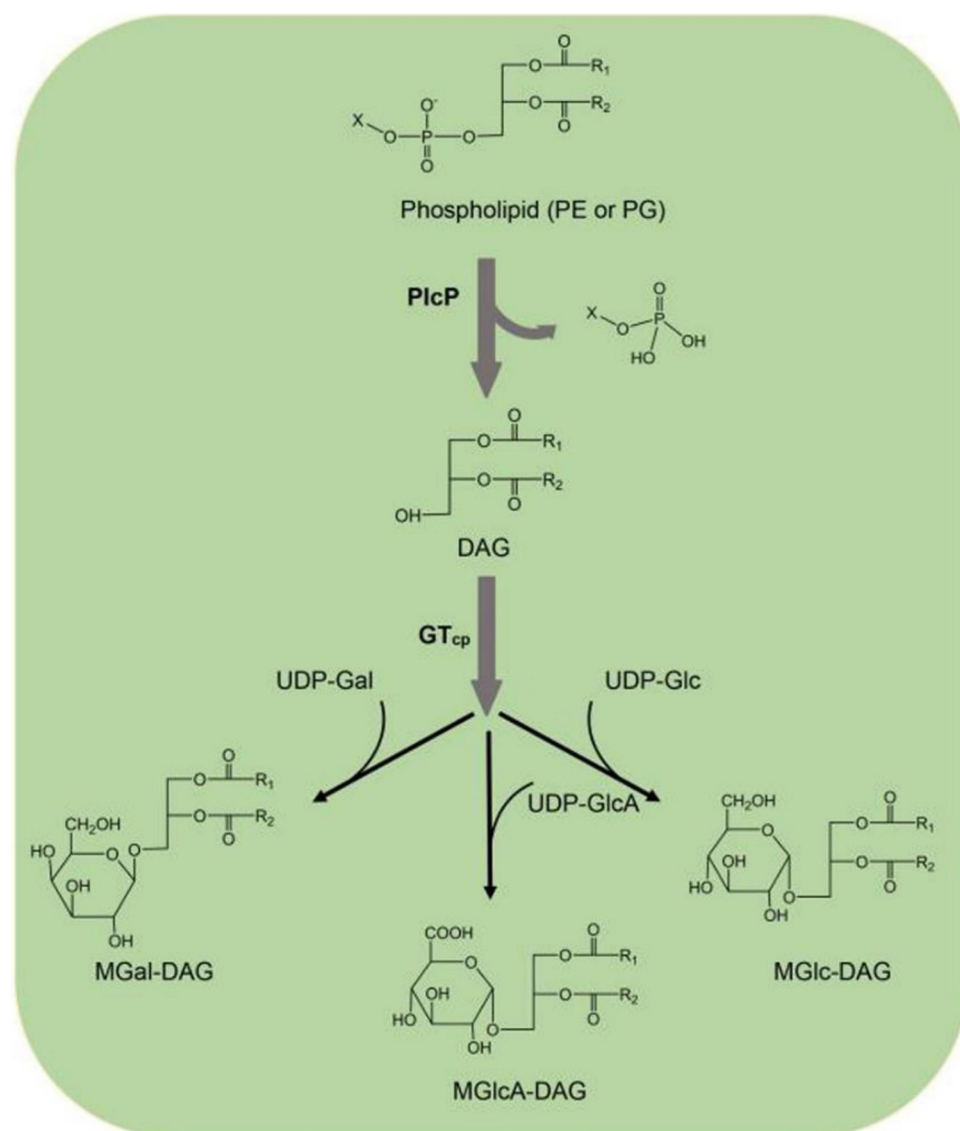
In MshA, the hydrophobic residue Leu76 helps to stabilize the dimer (21). This residue corresponds to Leu56 in GT<sub>cp</sub> (Fig. 1B). To investigate the role of Leu56 in GT<sub>cp</sub>, a site-directed mutant to Ala was made. The mutant L56A protein was loaded onto a Superdex 200 column to analyze its oligomeric state. The eluted peaks of a L56A mutant corresponded to the monomer of GT<sub>cp</sub> (38 kDa) according to their elution volumes (Fig. 2B). No difference was observed between the wild-type profile and the other mutants during protein purification (data not shown). The mutant L56A showed approximately 85% of wild-type GT<sub>cp</sub> activity. Structural studies revealed that most of the residues involved in oligomerization are conserved with related GTs, and they appear to be primarily hydrophobic and aromatic residues that form an extensive hydrophobic interface between the monomers (45). These results demonstrated that residue Leu56 is essential for the stable dimerization of the protein but does not play a direct role in the catalytic reaction of GT<sub>cp</sub>.

To conclude, our data show that the activity of purified GT<sub>cp</sub> from the marine bacterium “*Candidatus Pelagibacter* sp.” HTCC7211 is sufficient for the synthesis of several glycolipids, including MGlc-DAG, MGal-DAG, and MGlcA-DAG. The ability to synthesize MGlc-DAG and MGlcA-DAG suggests that GT<sub>cp</sub> may play an important role in lipid remodeling in natural marine systems. GT<sub>cp</sub> and PlcP, a manganese-dependent metallophosphoesterase, are organized in an operon-like structure in numerous marine heterotrophic bacteria (16, 17). Upon phosphorus (P) deficiency, PlcP selectively degrades phospholipids, such as phosphatidylglycerol (PG) and phosphatidylethanolamine (PE), to DAG, which then serves as the substrate for the biosynthesis of these glycolipids by GT<sub>cp</sub> using UDP-Glc, UDP-Gal, or UDP-GlcA as sugar donors (Fig. 7). Both the phospholipid PG and the glycolipid MGlc-DAG are anionic under physiological conditions, and it is likely that they could be interchangeable while maintaining the desirable biophysical properties of the membrane. Indeed, this has also been documented in the SAR11 strain HTCC7211 (16, 17). Similarly, substitution of PG by the anionic sulfur-containing glycolipid SQDG has also been shown for marine cyanobacteria and phytoplankton (46). Together, our work thus points to the important role of glycosyltransferases as key enzymes in the synthesis of glycolipids in marine bacteria.

## MATERIALS AND METHODS

**General materials and microorganisms.** We purchased UDP-glucose (UDP-Glc), UDP-galactose (UDP-Gal), UDP-glucuronic acid (UDP-GlcA), UDP-xylose, UDP-rhamnose, UDP-mannose, UDP-fructose, and diacylglycerols (DAGs) from Sigma-Aldrich Co. (St. Louis, MO, USA). Nickel column and Superdex 200 gel filtration columns were from GE Healthcare (Buckinghamshire, UK). All other chemicals were of the highest reagent grade and were obtained from Sangon (Shanghai, China). The *E. coli* strains JM109 for DNA manipulation and BL21-CodonPlus (DE3)-RIL for protein expression were obtained from TaKaRa Bio, Inc. (Dalian, China) and Stratagene (La Jolla, CA, USA), respectively.

**Cloning, expression, and purification of GT<sub>cp</sub>.** The gene GT<sub>cp</sub> from “*Candidatus Pelagibacter* sp.” HTCC7211 was codon optimized and chemically synthesized by Sangon (Shanghai, China). Several site-directed GT<sub>cp</sub> mutants (encoding mutations G16A, L56A, G82A, G85A, H104A, T162A, R163A, R190A, K195A, D256A, D256E, T257A, T257S, F258A, G259A, E264A, and W320A) were constructed using the overlapping PCR method with the common primers as for the wild-type GT<sub>cp</sub> and two site-specific primers for each mutant (Table 5). The genes were then cloned into the pET22b expression vector using NcoI and SalI restriction sites and transformed into the host *E. coli* BL21-CodonPlus (DE3)-RIL for gene expression. The transformed cells were grown in LB medium containing 100 mg/liter ampicillin and 34 mg/liter chloramphenicol at 37°C with shaking at 180 rpm. When the cultures reached an optical density at 600 nm (OD<sub>600</sub>) of 0.6, IPTG was added to a final concentration of 0.5 mM. After a further 4 h of growth at 37°C, the cells were harvested by centrifugation and lyophilized by vacuum freezing. The harvested cells were resuspended in buffer A (50 mM Tris-HCl [pH 7.9] and 50 mM NaCl), with 1% (wt/vol) of Triton X-100, and then disrupted by sonication. The cell mixture was then centrifuged at 12,000 × *g* for 30 min, and the soluble fraction was loaded onto a nickel column (GE Healthcare) pre-equilibrated with buffer A. The recombinant enzymes were eluted with elution buffer (20 mM Tris-HCl [pH 7.9], 500 mM NaCl, and 300 mM imidazole) and dialyzed overnight in buffer A to remove imidazole. For further purification, the enzymes were loaded on a Superdex 200 (16/60) gel filtration column (GE Healthcare), which was pre-equilibrated with buffer B (50 mM Tris-HCl [pH 7.9] and 200 mM NaCl). The fraction size was



**FIG 7** Proposed pathway of the synthesis of nonphosphorus glycosylglycerolipids through PlcP and GT<sub>cp</sub> in "*Candidatus Pelagibacter* sp." HTCC7211. PlcP converts phosphatidylglycerol (PG) or phosphatidylethanolamine (PE) to generate diacylglycerol (DAG), and GT<sub>cp</sub> can synthesize different glycosylglycerolipids MGal-DAG, MGlc-DAG, and MGlcA-DAG with UDP-Glc, UDP-Gal, or UDP-GlcA as sugar donors and DAG as the acceptor.

0.5 ml and the flow rate was 0.5 ml/min. The peak fractions were collected, concentrated, and analyzed by SDS-PAGE (12% polyacrylamide). The protein concentration was determined using the Bradford method. The purified protein was stored in buffer A containing 25% glycerol at  $-80^{\circ}\text{C}$ . SDS-PAGE gels and circular dichroism (CD) spectroscopy analyses of the purified protein and the mutants are shown in Fig. S2 in the supplemental material.

**Bioinformatics and homology modeling.** A putative GT gene encoding GT<sub>cp</sub> was identified in the genome of "*Candidatus Pelagibacter* sp." HTCC7211 (GenBank accession no. [WP\\_008545403.1](https://www.ncbi.nlm.nih.gov/nuclseq/CP008545.1)). ClustalW2 software (<http://www.ebi.ac.uk/Tools/clustalw2/index.html>) was used for multiple sequence alignment analysis of GT<sub>cp</sub> (47). For the phylogenetic analysis, we used the neighbor-joining method and Molecular Evolutionary Genetic Analysis (MEGA) software (v7.1) (48). The three-dimensional model structure of GT<sub>cp</sub> was generated using tools at the Phyre 2 protein modeling server (49) and the crystal structure of MshA (Protein Data Bank [PDB] entry [3C4Q](https://www.rcsb.org/entry/3C4Q)) from *C. glutamicum* as the template. Docking of UDP-Glc/UDP-Glc/UDP-GlcA with GT<sub>cp</sub> was predicted using the Flexible Docking module in Accelrys Discovery studio (50, 51), and the protein model was imported into Flare (v3.0, Cresset) for docking the DAG substrate and firstly energy minimized with 2,000 iterations with a cut off of 0.200 kcal/mol/Å. The DAG lipid was imported as a ligand and energy minimized in Flare before being docked into the active site and the best scoring pose selected. Thereafter we used the DAG-docked ligand with the model as the basis for the follow-up *in silico* docking of UDP-sugars in the presence of the DAG.

**TABLE 5** Primers used for PCR amplification in this study

Primer sequence (5'–3') <sup>a</sup>		
Vector	Upper	Lower
GT <sub>cp</sub>	GCCGCATATGAAAATTTAATCGTAAC (NcoI)	GGCGTCGACATTAGGTGATATTAAG (Sall)
G16A	CCACTTGTGAATGCTGTAGTTCGAAC	GTTCTGAACACAGCATTACACAAGTGG
L56A	GAAATTAGAGCATCATTAAATGTTT	CAAAACATTTAATGATGCTCTAATTC
G82A	CATATTGCAACAGAGGCACCTCTTGG	CCAAGAGGTGCTCTGTTGCAATATG
G85A	GGGACCTCTTGCTTTTATGGCAAG	CTTGCCATAAAAGCAAGAGGTCCC
H104A	CAACAAGTTTTGCTACAAGATTG	CAAATCTTGTAAGCAAACTTGTG
T162A	GGTTACATGGGCTAGGGGAGGTAATC	CCATGATTACCTCCCCTAGCCCATGTAAC
R163A	GGTTACATGGACTAGCGGAGGTAATCATGG	CACCATGATTACCTCCGCTAGTCCATGTAAC
R190A	GGATATACGTCGGTGCAAGTTCAGTTG	CAACTGCAACTGCACCGACGTATATCC
K195A	GCAGTTGAAGCAAATATTAAAGC	GCTTTAATATTGCTTCAACTGC
D256A	CCCTAGCAAAACGAATACTTTTGG	CCAAAAGTATGCGTTTTGCTAGGG
D256E	CCCTAGCAAAACGAATACTTTTGG	CCAAAAGTATTCGTTTTGCTAGGG
T257A	CCCTAGCAAAACCGATGCTTTTGGTATTG	CAATACCAAAAAGCATCGTTTTGCTAGGG
T257S	CCCTAGCAAAACCGATAGTTTGGTATTG	CAATACCAAAAAGCATCGTTTTGCTAGGG
F258A	CCGATACTGCTGGTATTGTGTTTTGG	CCAAAACCAACAATACCAGCAGTATCGG
G259A	CCGATACTTTTGCTATTGTGTTTTGGAG	CTCCAAAACCAACAATAGCAAAAGTATCGG
E264A	GTGGTTTTTGAGTCTTTAAGTTGTGG	CCACAACCTAAAGACTCCAAAACAC
W320A	GCTAAAAATATAGTTCGGAAGAAAC	CCTTGCTGTTCTTCCGAAGTATTTTTAG

<sup>a</sup>The restriction sites are underlined, and mutations are shown in boldface.

**Enzyme activity assay.** The enzymatic activity of GT<sub>cp</sub> was measured using 0.1 mM UDP-Glc (or UDP-Gal or UDP-GlcA) and 0.1 mM DAG as the substrate and 2.0 μM purified enzyme in 10 mM Tricine-KOH [pH 8.5] and 2 mM dithiothreitol (DTT). The resulting mixture (500 μl) was incubated at 35°C for 60 min with constant shaking at 200 rpm. The products (glycoglycerolipids and DAGs) were extracted using the Floch method with methanol-chloroform-water at a ratio of 1:2:0.6 (vol/vol/vol). The lipid extract was dried under nitrogen conditions at room temperature. The dried lipids were resuspended in acetonitrile and ammonium acetate (10 mM; pH 9.2) at a ratio of 95:5 (vol/vol) and analyzed by LC-MS. One unit of enzymatic activity was defined as the amount of enzyme required to catalyze the conversion of 1 μmol DAG per min under the standard conditions. The measurements were corrected for background hydrolysis in the absence of the enzyme. The  $K_m$  and  $V_{max}$  values were calculated using Hanes-Wolff plots with various concentrations of substrate (0.02 to 1.0 mM) and three replicates.

**Lipid analysis by TLC and LC-MS.** The GT<sub>cp</sub>-synthesized glycoglycerolipids were analyzed by TLC using a Camag Automatic TLC Sampler III (Camag, Muttens, Switzerland) for spotting. Glycoglycerolipids were separated on silica gel 60 (Merck, Darmstadt, Germany) with chloroform-methanol-water (65:35:4 vol/vol) and stained with sulfuric acid-methanol-water (45:45:10 vol/vol) for visualization. The resulting solutions were further analyzed by high-performance liquid chromatography using a 1290 Infinity II ultraperformance liquid chromatography (UPLC) instrument (Agilent Corp., Santa Clara, CA, USA) coupled with an Triple Quad 5500 (AB Sciex, Framingham, MA, USA) equipped with an electrospray-ion (ESI) detector. A BEH Amide XP column (2.5-μm inner diameter; 3 mm by 150 mm; Waters, Milford, MA, USA) was used for chromatographic separation. The mobile phase consisted of acetonitrile (solvent A) and 10 mM ammonium acetate (pH 9.2) (solvent B). The column was equilibrated for 10 min with 95% A:5% B prior to sample injection. The separation was conducted using a stepwise gradient starting from 95% A:5% B to 70% A:30% B after 15 min with a constant flow rate of 150 μl min<sup>-1</sup>. Mass spectrometric analysis was performed in the ESI positive ion mode with the ion spray voltage at 3,500 V and temperature at 350°C. The nebulizer gas and heater gas were set at 40 lb/in<sup>2</sup>. The analytical data were processed by Analyst software (v1.6.3).

**Inductively coupled plasma-mass spectrometry.** The metal content of GT<sub>cp</sub> was measured by using an ICP-MS (Agilent Technologies 7900 ICP-MS). The standards for calibration were freshly prepared by diluting Ca, Mg, Mn, Zn, and S stock solution (at 1,000 mg · liter<sup>-1</sup>; Sigma-Aldrich, Saint Louis, MO, USA) with 1% (vol/vol) nitric acid with concentrations from 0.1 to 2.0 mg · liter<sup>-1</sup> for Ca, Mg, Mn, and Zn and from 1 to 25 mg · liter<sup>-1</sup> for S. About 3.0 mg protein was digested in 1% (vol/vol) nitric acid matrix for metal analyses. The content of S was quantified in order to determine the protein concentration. The contents of Ca, Mg, Mn, Zn, and S were measured using the emission lines of 396.847 nm (Ca), 280.270 nm (Mg), 259.373 nm (Mn), 213.856 nm (Zn), and 180.669 nm (S), respectively.

**Nuclear magnetic resonance spectroscopy (NMR) spectroscopy.** NMR spectroscopy experiments were carried out in CDCl<sub>3</sub> with tetramethylsilane as an internal standard. <sup>1</sup>H, <sup>13</sup>C, distortionless enhancement by polarization transfer (DEPT), correlation spectroscopy (COSY), heteronuclear single quantum coherence (HSQC), HMBC, and rotating-frame nuclear Overhauser effect spectroscopy (ROESY) experiments were recorded at 298 K with a 600-MHz spectrometer (Bruker Avance 600; Bruker). Bruker standard software Topspin 3.2 was applied to acquire and process all the spectra data. COSY and ROESY experiments were recorded using data sets (t1 by t2) of 2,048 by 256 points, COSY with 4, and ROESY with 16 scans.

**Characterization of recombinant GT<sub>cp</sub>.** The optimum temperature of GT<sub>cp</sub> was measured in Tricine-KOH buffer (pH 8.5) in the range of 10 to 50°C. The buffer was adjusted to pH 8.5 for each of the assayed temperatures. The activation energy of the cleavage reaction was calculated using the logarithmic form of the Arrhenius equation, as follows:  $\ln k = \ln k_0 - E_a/R \cdot T$ . The effect of pH on enzymatic activity was tested at 35°C for pH values in the range of 7.0 to 11.0. The following buffers were used: sodium phosphate (pH 6.0 to 7.5), Tricine-KOH (pH 7.5 to 9.5), and *N*-cyclohexyl-3-aminopropanesulfonic acid (CAPS; pH 9.5 to 11.0). The thermostability of the purified GT<sub>cp</sub> was examined by incubating the enzyme in 50 mM Tricine-KOH buffer (pH 8.5) at three different temperatures (30, 40, and 50°C). Samples (80 μl) of the enzyme were collected after incubation periods of 30, 60, 90, and 120 min at each temperature, and the residual activity of each sample was assayed under standard conditions. The enzymatic activity of GT<sub>cp</sub> was also measured in the presence of various metal salts (MnCl<sub>2</sub>, ZnCl<sub>2</sub>, MgCl<sub>2</sub>, CaCl<sub>2</sub>, BaCl<sub>2</sub>, CdCl<sub>2</sub>, HgCl<sub>2</sub>, CuCl<sub>2</sub>, FeSO<sub>4</sub>, NiSO<sub>4</sub>, and CoCl<sub>2</sub>) at 5 mM or 5 mM EDTA. To determine the salt stability of the enzyme, 0 to 4 M NaCl (final concentration) was added to the reaction mixture, and enzyme activity was determined using optimum conditions.

**Data availability.** The authors confirm that the data supporting the findings of this study are available within the manuscript and its supplemental materials.

## SUPPLEMENTAL MATERIAL

Supplemental material is available online only.

**SUPPLEMENTAL FILE 1**, PDF file, 0.4 MB.

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